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(54) **Extraction of vegetable oils**

(57) Vegetable oils and protein of high quality can be extracted from oil seeds by contacting the seeds with water in the presence of at least one added enzyme, and separating the resulting reaction mixture into a solid protein-containing fraction, an oil-fraction and an aqueous fraction. Advantageously a water-immiscible solvent e.g. hexane is present during the extraction, or can be added to assist the separation stage.

SPECIFICATION

Extraction of vegetable oils

5 This invention relates to the extraction of oils from seeds. More particularly, this invention relates to the extraction of oils and proteins from oil seeds by an enzymic process.

Oil seeds, e.g. soybeans, peanuts, cotton seeds, 10 sunflower and rape seeds constitute the major source of edible oils. As much as 70% of world production of fats or oils is provided from vegetable sources, more than 20% coming from soybeans. In addition to providing oils, these seeds constitute also an increasingly-important source of protein.

Two main types of process are commonly employed for the commercial extraction of oils from oil seeds. These are (a) mechanical expression, and (b) solvent extraction. Both types of process tend to 20 denature the proteins occurring in the oil seeds, thereby lowering the solubility of the protein extracted, and restricting its utility in food products. There has been a search for improved extraction processes, but this search has been concentrated on aqueous extraction processes, involving centrifugal 25 separation of oil and protein phases.

The present invention provides a process for the extraction of vegetable oils and protein from oil seeds which comprises contacting oil seeds with water in the 30 presence of at least one added enzyme, and separating the resulting reaction mixture into a solid protein-containing fraction, an oil-fraction and an aqueous fraction.

In spite of the expanding interest in recent years in 35 the applications of biotechnology, there appears to have been no disclosure of such a process applied to the extraction of vegetable oils. Indeed, those enzymes occurring naturally in the oil seeds are usually inactivated before the extraction processes generally employed in industry. The purpose of this inactivation 40 step is to avoid decomposition processes leading to rancid and sometimes toxic products.

It has now been found that by carrying out a water extraction of oil seeds, usually at an elevated temperature, in the presence of one or more of a variety of 45 enzymes, it is possible to obtain good yields of the desired vegetable oil, and in addition to obtain a protein fraction which has undergone less denaturing than occurs in the processes conventionally used in industry. Moreover, the oil obtained is of high quality, 50 and does not generally require the elaborate purification commonly necessary to remove the degradation products obtained in conventional extraction processes.

The process of the invention of the invention even 55 makes it possible to produce oil from sources not normally considered as oil sources, e.g. melon seeds.

A variety of enzymes can be employed in the process of the invention. Indeed it is generally desirable to employ enzyme mixtures (from the same 60 or different sources) so that different enzymes may each perform their different functions leading to a more rapid and effective breakdown of the cellular structures in the oil seeds, and more effective release of the oils.

65 Among the types of enzyme that may be employed

are α -amylase, β -glucanase, neutral-proteinase, alkaline proteinase, acidic proteinase, pectolytic enzymes, cellulase, and hemicellulase. Specific enzymes that have been effectively employed in the process of the 70 invention are commercially available mixtures, e.g. a mixture of α -amylase, β -glucanase and n-proteinase from *B. subtilis*; a mixture of pectolytic enzymes and cellulases from *A. niger*; and hemicellulase from *A. niger*.

75 The enzymic extraction is generally carried out at ambient or at an elevated temperature, for instance 15 to 90°C, more preferably 45 to 70°C, the actual temperature being one at which the enzymes function well without being deactivated. The pH can also be 80 adjusted during the course of the reaction to a value suitable for the particular enzyme being used. Values in the range from 3 to 8 are generally suitable.

Separation of the oil fraction from the reaction product is assisted by the use of a solvent. This can be 85 any water-immiscible solvent whose volatility enables its easy recovery, e.g. hexane. This solvent can be present, if desired during the enzymic extraction, or can be added after or preferably before filtration. Such solvents are generally those used in known processes 90 for extraction of oils from seeds. They are, however, generally employed in lower quantities in this invention, because they fulfil a different function, so that there is less expenditure of energy on solvent recovery.

95 In accordance with one embodiment of the invention, the extraction process is carried out by size-reducing the oil seeds to a desired particle size. The resulting particles are contacted with water, suitably with stirring or other agitation, at a temperature of 100 from 15 to 90°C, e.g. from 45 to 70°C, for a suitable period e.g. of 1 to 6 hours. The enzyme or enzymes are present during this extraction step, e.g. in amounts up to 5% by weight, based on oil seeds. In general, increasing the amount of enzyme increases the speed 105 at which the oil is separated, but optimum amounts can be readily established.

Once the extraction has proceeded for a desired period, the water-immiscible solvent, e.g. hexane, can be added, and the resulting mixture is filtered. The 110 solids, thus-separated comprise a protein-containing fraction, and the further treatment of this fraction will be further described below. The filtrate comprises water, oil and solvent. Its pH can if desired be suitably adjusted, e.g. to 3.5 with an acid, such as citric acid, 115 and it is then centrifuged to give an insoluble fraction, an aqueous fraction and an oil/solvent fraction, the latter two fractions being separated by decantation. The desired vegetable oil is isolated by evaporating, and desirably recycling the solvent.

120 The insolubles from the centrifuging step can be combined with the solids from the filtration step, and purified by extraction with a solvent. The residue after removal of solvent constitutes a highly advantageous protein concentrate.

125 In an alternative and preferred embodiment, the solvent is added before or during the extraction step. The later steps of the process are otherwise the same as those of the process described above. This embodiment provides higher yields of oil.

130 The following Examples set out results obtained

when working on a laboratory scale. In these Examples, the following enzymes were employed:

Enzyme — C : a mixture of α -amylase, β -glucanase and n-proteinase from *B. subtilis* (Ceremix-Novo

5 Industri A/S)

Enzyme — P : a mixture of pectolytic enzymes and cellulases from *A. niger* (Pectinex-Swiss Ferment Company)

Enzyme — G : hemicellulase from *A. niger*

10 (Gamanase-Novo Industri A/S)

A1 *A. niger*—fungal β -glucanase

A2 *A. niger*—fungal hemicellulase

A3 *A. niger*—fungal pectolytic enzyme complex

T *Trichoderma viridii* —fungal

15 EXAMPLE 1

Rapeseed was mechanically ground using a coffee mill.

25g of ground meal was suspended in 100 ml of distilled water in a conical flask.

20 0.25 ml of enzyme C was added, and the pH was adjusted to 6.0 with M/l NaOH. The flask and contents were incubated in a waterbath and stirred electrically, the pH being maintained at 6.0.

The following temperature programme was used:—

60 minutes at 50°C

Temperature allowed to rise to 63°C by adjusting thermostat.

120 minutes at 63°C

30 Temperature allowed to rise to 80°C

13 minutes at 80°C

On completion of this sequence, the flask and contents were cooled to room temperature.

The contents of the flask were coarse-filtered through a cheese cloth (to remove solid debris then washed with excess hexane. The pH of the filtrate was lowered to 3.5 (with M/l citric acid) and it was centrifuged to give 3 fractions:

1. Oil/solvent layer

40 2. Aqueous layer

3. Insoluble residues

The top layer was decanted and the aqueous layer was separated from insoluble by filtration through paper.

45 Insolubles were pooled, dried, ether extracted (to remove any residual oil) and dried.

The yield of oil was calculated by determining the amount of residual oil (by weight, x) after evaporating the solvent from fraction 1 and expressing this value as a fraction of the total weight of the raw material taken.

$$\text{i.e. yield} = \frac{x}{25} \cdot 100\%$$

The amount of oil extracted (average of 5 separate experimental runs) was found to be 9.3%

The amount of extractable oil in rapeseed and 55 soybeans was estimated by Soxhlet extraction of samples of the ground materials with hexane, followed by evaporation of the solvent from the extracted material. The yields of oils were as follows:—

60 Rapeseed: 30.5%

Soybean: 19.8%

(These values therefore represent ideal control values — the upper limits sought in the subsequent experiments).

EXAMPLE 2

The same experiment procedure was used as in Example 1, except that double the amount of enzyme C was used (i.e. 2% v/w basis on seed). The amount of oil extracted (average of 5 separate experimental runs) was 12.8%.

EXAMPLE 3

The same experimental procedure was used as in Example 1, except that three times the amount of enzyme was used (i.e. 3% v/w basis on seed). The amount of oil extracted (average of 5 separate experimental runs) was 13.3%.

EXAMPLE 4

The same experimental procedure was used as in Examples 1-3, except that ground soybean was used as source material instead of rapeseed. The amount of oil extracted (each the average of 5 separate experimental runs) was as follows:—

Enzyme concn, % v/w basis on bean	1	2	3
Oil Yield % (calc. as in Ex1)	7.1	8.8	9.2

EXAMPLE 5

The same experimental procedure was used as in Example 1, with the modification that in addition to 100 ml of distilled water and enzyme solution, 50 ml of hexane was also added to the milled rapeseed to extract the liberated oil. Hence liberation of oil from seed is effected by simultaneous enzymatic hydrolysis and product removal by solvent extraction. A condenser was employed in order to prevent loss of solvent during the temperature stepped hydrolysis.

The amount of oil extracted (average of 5 separate experimental runs) was 16.7%.

EXAMPLE 6

The same experimental procedure was used as in Example 5 but 2 and 3 times the amount of enzyme C was used (i.e. 2% and 3% v/w basis on seed). The amount of oil extracted (average of 5 experimental runs) was:—

Enzyme concn, % v/w basis on seed	2	3
Oil Yield % (Calc as in Ex1)	19.8	20.0

EXAMPLE 7

The same experimental procedure was used as in Example 5 and 6, except that ground soybean was used as source material instead of rapeseed. The amount of oil extracted (average of 5 separate experimental runs) was as follows:—

Enzyme concn, % v/w basis on bean	1	2	3
Oil yield, % v/w	10.6	12.4	12.6

EXAMPLE 8

The same experimental procedure was used as in Example 5, 6 and 7 (i.e. both rapeseed and soybeans) except that enzyme P was used in place of enzyme C, and a pH of between 4.5 and 5.0 was maintained, instead of a value of 6.0. The amount of oils extracted (each an average of 5 separate experimental runs for both types of materials and for the various enzyme

levels) were as follows:—

Enzyme Conc.	% v/w basis	1	2	3
Oil Yield,	% w/w rapeseed	18.5	21.4	21.5
	soybean	14.2	16.4	16.6

EXAMPLE 9

The same experimental procedure was used as in Example 8, except that enzyme G was used. The amounts of oils extracted (average as in Example 8) were as follows:—

Enzyme concn.	% v/w basis	1	2	3
Oil Yield	% w/w rapeseed	19.1	22.0	22.1
	soybean	15.2	17.5	17.8

EXAMPLE 10 (CONTROL)

The same experimental procedure was used as in Example 5, except that (as a control) no enzymes were used. The amount of oils extracted (each an average of 3 separate experimental runs) were as follows:—

	Rapeseed	Soybean
Oil Yield, % w/w	3.8	2.7

EXAMPLE 11 (PROTEIN CONTENT)

The crude protein content of the residual insolubles obtained after oil extractions was estimated by the standard Kjeldahl method (NX6.25). The results were as follows (average of 3 separate experiments):—

	% crude protein in protein concentrate
Rapeseed	27.1
Soybean	33.6

EXAMPLE 12 (OIL QUALITY)

An analysis of the quality of the rapeseed oils obtained was made by the standard method as described in Pearson (The Chemical Analysis of Foods, 7th Edn, 1976, pub: by Churchill-Livingstone, Edinburgh) and the results compared with ideal figures as recommended by codex standards:—

Parameter	Unit	Results	Codex Standard
Iodine value	—	98	94 - 120
Saponification value	mgKOH/g	174	168 - 181
Unsaponifiable matter	g/Kg	17.8	max 20
Peroxide value	meq/Kg	9.62	max 10
Refractive index	(at 4°C)	1.467	1.465-1.469
Acid value	mgKOH/g	0.38	max - 0.6

EXAMPLE 13

(for Comparison, Preparation of Protein Concentrate)

Dried melon seeds were ground in a similar manner as in Example 1.

The resultant flour was weighed and homogenised to produce a slurry. This was treated with acid to lower the pH to 3.5 and diluted with water to give a 1:3.5-4 ratio of seed flour to liquid.

This was then centrifuged, with the result that the extract separated into 3 layers: solids at the bottom, an aqueous layer in the middle and an oily layer floating on the surface.

The liquid layers were decanted and the sludge was recentrifuged to give a solid and an oily layer floating on top.

Liquid layers were recentrifuged to sediment suspended solids.

Solids were then combined, dried and extracted with diethyl ether and dried to give a protein concentrate fraction. Oil was collected by decantation and evaporation of solvent, and estimated by weight.

The following results were obtained:—

	g	% approx:
Mass of melon seed flour	352	
Mass of water used to extract flour	1256	
Net mass of oil floating	104	30
Mass of protein concentrate (wet)	250	
Mass of protein concentrate (dry)	170	
Mass of protein concentrate after Et ₂ O Extn	136	40
Oil removed after aqueous processing		6.5
Oil removed after Et ₂ O Extn		30
Oil and Solubles		60

* Protein was estimated as Nx5.7 - nitrogen determined by macro-kjeldahl method.

EXAMPLE 14

12g of protein concentrate prepared as in Example 13 was suspended in 20 ml of water in a flask. The pH was adjusted to 8.0 with NaOH and the suspension was heated to 50°C in a water bath.

16.8m Anson units of alkaline proteinase (produced as a culture filtrate by fermentation of *Bacillus licheniformis*) was added to the flask, and the whole was diluted to constant weight with water.

The mixture was incubated at 50°C for 4 hours, whilst the pH was maintained at 8.0 by addition of M NaOH from a burette.

The flask was then reweighed (to allow for evaporation corrections) and the pH was lowered to 4.5 with citric acid.

The temperature was raised to 80°C for 5 minutes to inactivate the enzyme.

The contents of the flask was then analysed and gave the following results:—

	%
Solubilised protein	35
Protein hydrolysate	2.8
"Extra" oil released	16

EXAMPLE 15

A series of experiments similar to that outlined in Example 14 were run.

The substrate used was 8% v/w protein isolate and incubations were again carried out at 50°C but with pH maintained at 4.5. The following enzymes were used in addition to the alkaline proteinase:—

A1 *Aspergillus niger*-fungal β -glucanase
T *Trichoderma viridi*-fungal cellulase
A2 *Aspergillus niger*-fungal hemicellulase
A3 *Aspergillus niger*-fungal pectolytic enzyme complex

The following results were obtained:—

Enzyme Type Used	Solubilised Protein %	Hydrolysed Protein %	"Extra" Oil Recovered %
A1	38	3.0	14
T	43	3.4	14
A2	47	3.5	19
A3	48	3.8	23

EXAMPLE 16

An experiment similar to Example 14 was run, using whole flour instead of protein extract, and a mixture of α -amylase, β -glucanase, and n-proteinase enzymes produced from *Bacillus subtilis* was used to supplement the α -proteinase from *B. licheniformis*. The results obtained were as follows:—

	%
Solubilised protein	25
Hydrolysed protein	4.2
Oil released	18

Examples 13 to 16 illustrate that oil can be released from alternative oil seeds by enzymatic hydrolysis of seedmeal or the residual oil bound to a protein isolate produced from the seedmeal in which indigenous oil had been removed by aqueous floatation or solvent extraction.

CLAIMS

1. A process for the extraction of vegetable oils and protein from oil seeds which comprises contacting oil seeds with water in the presence of at least one added enzyme, and separating the resulting reaction mixture into a solid protein-containing fraction, an oil-fraction and an aqueous fraction.
2. A process according to Claim 1 wherein the reaction product is filtered to separate the protein-containing fraction, a water-immiscible solvent is added, and the solvent and vegetable oil are separated from the aqueous fraction.
3. A process according to Claim 1 wherein the oil seeds are contacted with water in the presence of the enzyme and of a water-immiscible solvent, and the reaction product is filtered to separate the protein-containing fraction, and the resulting filtrate is separated into a solvent-oil fraction and an aqueous fraction.
4. A process according to any of the preceding Claims wherein the oil seeds are contacted with water and the added enzyme at an elevated temperature.